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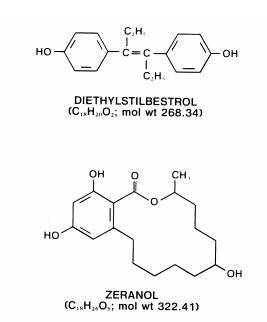
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A. INTRODUCTION

1. Theory

This procedure uses a three-phase solvent extraction of tissue using an aqueous buffer, acetonitrile, dichloromethane, and hexane to effectively remove the majority of the triglycerides and highly non-polar materials prior to loading on the column. This step prevents the overloading of the column, which would reduce recoveries, and simultaneously fractionates compounds into three polarity classes. An extraction cartridge containing a strongly basic anion exchange resin with guaternary amine groups in the chloride form is used to extract the DES and zeranol. The phenolate anion of these compounds has a high affinity for the chloride form of the resin at pH values above 12.0. The acetonitrile extract of the tissue is concentrated, made alkaline, and the analyte is exchanged to the resin. Several solvent washes are performed to remove interfering substances. A 5% aqueous acetic acid wash allows for the elution of a large quantity of interfering anionic materials previously exchanged to the resin. The column is then eluted with a solution of limited solvent strength (25% agueous methanol) to remove residual acetic acid and additional matrix material prior to the final elution with methanol. The procedure is amendable to automation using the four-solvent Dupont Prep I centrifugal column elution device, which significantly increases sample throughput.

2. Structure



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3. Applicability

This method is applicable for DES and Zeranol in ovine and bovine liver and muscle at levels \geq 0.5 ppb DES and 1.0 ppb Zeranol.

B. EQUIPMENT

Note: Equivalent apparatus or instrumentation may be substituted for any of the following.

- 1. Apparatus
 - a. Polypropylene centrifuge tubes 50 mL, screw-cap Falcon.
 - b. Homogenizer Tekmar SDT Tissuemizer with 1810 motor and SDT-100EN shaft and generator and TR-10 high torque speed control.
 - c. Centrifuge IEC PR-6000, equipped with 12-place angle rotor, and capable of achieving a G force of \ge 2500.
 - d. Pasteur disposable pipets.
 - e. Folded Filter Paper 125 mm, #2, Whatman.
 - f. Repipet dispensers 10 mL, Labindustries, Inc.
 - g. Low temperature indicator Precision Scientific, Freas Model 815, or constant temperature water bath (35 37 °C).
 - h. Type AS anion exchange cartridge.
 - i. Plastic recovery cups.
 - j. Prep I automated sample processor Dupont Chemical Co., catalog #60000-908.
 - k. Reacti-Therm dry block heater with custom-drilled heat blocks, Pierce Chemical Co.
 - I. Reacti-therm evaporator unit attachment Pierce Chemical Co.
 - m. Fisher sample concentrator Model 190, custom-drilled to accept 20 mL. scintillation vials, catalog #09-549.
 - n. Vortex mixer.
 - o. Tapered 1 mL auto sampler vials 1 CWVC combo-pack and 8 pep-6 polyethylene plug, Chemical Research Supplies.
 - p. Eppendorf pipets.
 - q. Disposable pipet 10 mL.
 - r. pH meter Orion Model 601 A.

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- s. Syringe 100 µL.
- t. Rotorack or Eberbach shaker (optional).
- u. Balance MT5, capable of reading to + 0.001 mg, Mettler-Toledo, Inc, Columbus, OH.

2. Instrumentation

 $\rm GC/MS$ - Hewlett-Packard, 5890 GC, 5970A series mass selective detector and 98258 computer.

C. REAGENTS AND SOLUTIONS

Note: Equivalent reagents and solutions may be substituted for any of the below.

1. Reagents

- a. Acetonitrile Burdick & Jackson, high purity solvent grade.
- b. Hexane Burdick & Jackson, high purity solvent grade.
- c. Dichloromethane Burdick & Jackson, high purity solvent grade.
- d. Methanol Burdick & Jackson, high purity solvent grade.
- e. Ethyl acetate Burdick & Jackson, high purity solvent grade.
- f. Isopropyl alcohol Burdick & Jackson, high purity solvent grade.
- g. Water distilled, deionized.
- h. Acetic acid, glacial USP #9522, Canlab Chemicals.
- i. Derivatizing agents BSTFA, Pierce Chemical Co., #38830; TMSI, Pierce Chemical Co., #88623.
- j. Sodium hydroxide reagent grade pellets #3722-1, J. T. Baker Chemical Co.
- k. 2N Sodium acetate trihydrate Fisher Certified Reagent, #S209B, Fisher Chemical Co.
- I. β-Glucuronidase Type HP-2S Cat No G7770, Sigma Chemical Corp., St. Louis, MO. Activity approximately 100,000 units/mL.

2. Solutions

a. 2N Sodium hydroxide:

80 g/L in distilled, deionized water.

b. 1:1 Isopropanol:methanol:

100 mL of each, measured separately and mixed.

c. 5% Aqueous acetic acid:

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5 mL glacial acetic acid made up to 100 mL with distilled, deionized water.

d. 25% Aqueous methanol:

25 parts methanol; 75 parts distilled, deionized water; measured separately and mixed.

e. 0.04M Sodium acetate trihydrate buffer:

5.44 g sodium acetate dissolved in 1 L distilled, deionized water.

D. STANDARDS

Note: Equivalent standards and solutions may be substituted for any of the following.

1. Source

- a. Zearalane zeranol assay internal standard, Schering-Plough Animal Health, Union, NJ.
- b. Zeranol (alpha-zearalanol) #Z 0292, 5 mg, Sigma-Aldrich Chemical Company, St. Louis, MO. (Five bottles are needed to make up the primary standard as directed below.
- c. D₈DES DES internal standard (diethyl-1,1,1',1 '-d4-stilbestrol-3,3',5,5'-d4), Cambridge Isotope Laboratories, Andover, MA.
- d. DES 99% purity, #21894-4, Aldrich Chemical Co., Milwaukee, WI.
- e. Diethylstilbestrol monoglucuronide (DES-MG) #1 6444-5, Aldrich Chemical Co., Milwaukee, WI. Each ampoule contains the equivalent of 1 mg of free DES.
- 2. Preparation of Standards
 - a. DES, D_8 DES, zearalane, zeranol.
 - i. Stock solutions:

Weigh 20.00 mg \pm 0.05 mg each of zearalane and zeranol and 10.00 mg \pm 0.05 mg each DES and D₈DES into separate 100 mL volumetric flasks. Dilute to volume with methanol (concentration 200 µg/mL for zearalane and zeranol and 100 µg/mL for DES and D₈DES).

ii. Working solutions:

Dispense 100 μ L of zearalane, zeranol, DES, and D₈DES stock solutions into separate 100 mL volumetric flasks. Dilute to volume with methanol (concentration 0.2 ng/ μ L for zearalane and zeranol, and 0.1 ng/ μ L for DES and D₈DES).

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b. Conjugated DES.

i. Stock solution (10 μ g/mL free DES):

Take 1 ampoule of DES-MG (equivalent to 1 mg free DES) and dissolve the contents with methanol in a 100 mL volumetric flask.

ii. Working solution (0.1 ng/ μ L free DES):

Dispense 1 mL of stock solution into a 100 mL volumetric flask. Dilute to volume with methanol.

Note: Allow standards to equilibrate to room temperature before use.

- 3. Storage Conditions: All standards should be kept frozen at < -10 °C if not in use.
- 4. Shelf Life Stability
 - a. Stock solutions: One year.
 - b. Working standards: Up to 6 months.

E. SAMPLE PREPARATION

Sample tissue should be thawed but cold before processing. Cut tissue pieces from various locations, avoiding fat. Place liver tissue in a blender and muscle tissue in a *Robocoupe*[®] or grinder. Process tissue just long enough to produce a homogenous blend without warming the tissue. Place samples in a freezer after preparation.

F. ANALYTICAL PROCEDURE

Note: All tissues should be kept frozen at all times until assayed.

- 1. Sample Extraction
 - a. Weigh 5.0 g + 0.1 g blended liver or ground, homogenized muscle into 50 mL polypropylene centrifuge tube.
 - b. Fortify each sample and control sample with 50 μ L (concentration 0.2 ng/ μ L) of zearalane and 50 μ L (concentration 0.1 ng/ μ L) of D₈DES internal standard (equivalent to 2.0 and 1.0 ppb) respectively.

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- c. Prepare controls as follows:
 - i. For screening sets

Weigh 2 tissue blanks and prepare duplicate recovery samples by adding zeranol (concentration 0.2 ng/ μ L), and DES-MG (concentration 0.1 ng/ μ L) as follows:

DES (ppb)	Zeranol (ppb)	µL of DES fortification solution added	μL of Zeranol fortification solution added
0.50	1.00	25.0	25.0

 For quantitative sets, weigh 4 tissue blanks and add zeranol (concentration 0.2 ng/μL), and DES-MG (concentration 0.1 ng/μL) as follows:

centration (ppb)	μL of DES fortification solution added	μL of Zeranol fortification solution added
Zeranol		
0.00	0.0	0.0
1.00	25.0	25.0
2.00	50.0	50.0
4.00	100.0	100.0
	0.00 1.00 2.00	fortification solution addedZeranol0.000.01.0025.02.0050.0

- d. With repipet, add 11 mL of 0.04 M sodium acetate buffer.
- e. Homogenize for 1 minute, using Tissuemizer.
- f. Adjust pH to 4.25 4.75 with glacial acetic acid. Add 5 drops of glacial acetic acid with a Pasteur pipet, vortex mix, and check pH (5 8 drops are usually required).
- g. Add 100 μ L ß-glucuronidase to each sample (approximately 10,000 units). Cap tube and vortex mix thoroughly for about 10 seconds.
- h. Incubate overnight (16 -18 hours) at 37 °C or place overnight in a constant temperature water bath set at 35 37 °C.
- i. After incubation, add 16.0 mL acetonitrile to each tube, using a repipet, and shake moderately by hand or in a Rotorack or Eberbach shaker for about 5

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	minutes.	•		
j.		es at a minimum G force of 2450	a for 5 minutes.	
J.	o 1	.118 x 10^{-5})(radius of rotation in	•	
		= distance from center shaft to th		
k.	Filter the superna polypropylene ce	tant through fluted filter paper int ntrifuge tube.	to a 50 mL screw-cap	
	Note: If a small ar will not interfere v	mount of floating debris is transfe vith the test.	erred with the supernatant, it	
I.	Add 2 mL dichlore	omethane and 8 mL hexane to th	e supernatant.	
m.	Shake moderatel	y by hand for 1 minute (or use Ro	otorack or Eberbach shaker).	
n.	Centrifuge at a m	inimum G force of 1060 for 2 mir	nutes.	
0.		ayer (acetonitrile) to a clean 20 m bet, without removing any of the o		
p.	With repipet, add another 4 mL of acetonitrile to the remaining two layers in the 50 mL centrifuge tube.			
q.	Shake moderately	y by hand for 1 minute (or use Ro	otorack or Eberbach shaker).	
r.	Centrifuge at a minimum G force of 1060 for 2 minutes.			
S.	Repeat step o above, recovering as much of the middle acetonitrile layer as possible using a 10 mL disposable pipet, without removing any of the other layers.			
	Stop point: Samp point.	les may be stored for up to 1 hou	ar at room temperature at this	
t.	or compressed ai	the combined acetonitrile layers r at 60 \pm 5 °C, using the Fisher M n of nitrogen, using the custom-d	10del 190 sample concentrator,	
	Note: All acetonit	rile must be removed. This takes	s 45 - 60 minutes.	
u.	With repipet, add minutes and vorte	2 mL 1:1 isopropanol:methanol a ex until dissolved.	and dissolve residue. Let set 5	
v.	With disposable p or swirl for 5 -10 s	vipet, add 1.5 mL 2N sodium hyd seconds.	roxide solution and vortex mix	
	Note: Column cle	anup must proceed at this point.		
W.	Place the cartridg	AS cartridges; decant liquid abo es on the inner ring of the rotor. e recovery cups on the outside ri	Place the plastic waste cups	
х.	Load sample solu	tions from step v above onto the	cartridge bed. Rinse the vials	

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with 1.0 mL of isopropanol:methanol and also add this to the cartridges.

- y. Close the lid of the Prep I and select Program 7. Advance the program to step 2 with the STEP ADVANCE switch and start the processor with the STEP ACTIVATE/HOLD switch. Allow the unit to run for 4 minutes and then stop it with the STEP ACTIVATE/HOLD switch. Open the lid and empty the waste cups and return them to the outer rotor.
- z. With repipet, add 4 mL methanol to each cartridge and repeat step y above.
- aa. Add the following solvents to solvent reservoirs 1, 2, 3, and 4 respectively.

i.	Wash solvent 1:	Place 18 mL of distilled deionized water in reservoir 1. This is equivalent to 1.4 mL per cartridge.
ii.	Wash solvent 2:	Place 33 mL of 5% aqueous acetic acid in reservoir 2. This is equivalent to 2.75 mL per cartridge.
iii.	Wash solvent 3:	Place 22 mL of 25% aqueous methanol in reservoir 3. This is equivalent to 1.75 mL per cartridge.
iv.	Elution solvent:	Place 33 mL \pm 3 mL of 100% methanol in reservoir 4. This is equivalent to 2.75 mL \pm 0.25 mL per cartridge.

Note: The volume of methanol elution solvent may require adjustment (\pm 3 mL) such that the final volume in step cc below is < 0.5 mL. The specific volume required for each Prep I unit in each laboratory can be predetermined by the analyst.

- bb. Select Program 7 and advance to step 3 with the STEP ADVANCE switch. Set the evaporation temperature to 55 °C (approximately setting 5). Start the program with the START switch.
- cc. After the program is finished, remove the plastic recovery cups and quantitatively transfer methanol eluate (0.1 0.5 mL) to a tapered 1 mL auto sampler vial using small MeOH rinses. Evaporate just to dryness under low nitrogen flow in a custom-drilled heat block set at 60 ± 5 °C. Note: If volume of eluate + rinses exceeds amount that can be safely concentrated without sample loss, use multiple partial transfer/evaporation cycles to effect concentration. The sample is ready for GC/MS analysis.

Note: Extracts must be stored in a freezer and should be analyzed as soon as possible but, if necessary, may be stored frozen up to a week without significant

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L	· ·	1		

loss.

- 2. Instrumental Settings
 - a. GC parameters

Note: Optimize instrumental conditions to obtain maximum sensitivity and adequate baseline resolution of peaks of interest, especially zeranol. The following table represents one optimization possibility.

i.	GC	Hewlett Packard 5890
ii.	Column	Crosslinked 100% methylsilicone HP special purpose 25.0 m, 0.2 mm i.d., 0.33 micron film thickness (16 m, 0.20 mm i.d., 0.5 micron film thickness has been used successfully)
iii.	Injector	Heated splitless, 260 °C minimum
lv.	Linear carrier gas flow	30 cm/sec, helium
٧.	Source Temperature	250 °C
vi.	Interface Temperature	250 °C
vii.	Temperature Program	Hold at 130 °C for 1 minute. Ramp at 20° /minute to 230 °C. Ramp at 5 °C/minute to 295 °C. Ramp at 5 °C/minute to 310 °C. Hold for 10 minutes.
MS	parameters.	

- i. Mass spectrometer Hewlett Packard 5970A-MSD ii. Ionization mode Electron impact iii. Electron energy 70 eV iv. Calibration Autotune Electron Multiplier Voltage Approximately 1300 V v.
- 3. **General Operation**

b.

- Reconstitute the dry residue with 10 µL of ethyl acetate and derivatize the a. samples with BSTFA plus 2% TMSI catalyst using an on-column derivatization technique.
- Draw 1.0 2.0 µL of sample into the syringe, followed by a 1 µL air space b.

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and 2 - 3 μ L of derivatization agent. (Wipe syringe needle with Kimwipe soaked with ethyl acetate before inserting into the derivatization agent).

c. Inject the entire contents (3 - 5 μL total) into the heated injection part of the GC operating in the splitless mode.

Note: 4 - 5 μ L of ethyl acetate solvent should be injected after every positive sample to prevent carry over interferences.

- 4 Sample Chromatogram See Section K for chromatograms.
- 5. Retention times

Approximate retention times
(min) with above conditions:

a.	cis DES	10.8
b.	trans DES	11.9
C.	Zearalane	14.9
d.	Zeranol	17.5

6. Ions Monitored

Analyte		lons (Dwells = 100 milliseconds)
a.	Zearalane	435
b.	Zeranol	538, 523, 433, 379
C.	D ₈ DES	420
d.	DES	412, 397, 383

G. CALCULATIONS

- 1. Procedure
 - a. Screening calculations
 - i. Report samples as positive if they meet the confirmation criteria specified in G. 2. a d.
 - ii. Report samples as negative if they fail to meet the confirmation criteria specified in G. 2. a d.

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b.	 Quantitative calculations Read tangential area counts or peak height for the selected ions from the integration report and tabulate. i. Obtain calibration curve values by ratioing the ion area of the analyte in question to the appropriate internal standard, as below: 			

zeranol =	Area counts or peak height for ions 538 + 523 + 433
	Area counts or peak height for ion 435

DES =	Area counts or peak height for ion 412, cis + trans
	Area counts of peak height for ion 420, cis + trans

ii. Using linear regression, construct a standard curve of DES or zeranol ion ratios vs. concentration of DES or zeranol.

The equation: y = mx + b. where: y = ion ratio x = conc. in ppb (µg/kg) m = slopeb = y intercept

The correlation coefficient should be > 0.995.

- iii. From the ion ratios, using the regression slope and intercept, calculate the concentration for each sample.
- 2. Criteria for Confirmation (Screening and Quantitative analyses)
 - Note: The set must include a tissue blank and three fortifications to bracket the estimated level of the suspect positive.
 - a. All monitored ions must be present for all isomers.
 - b. Blank tissue must show no interferences.

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c. Ion ratios must agree within 30% of those of the fortified standard.

Compounds	Ion Ratios
Zeranol	523/538 433/538 379/538
DES (cis + trans)	383/412 397/412

d. Retention time must agree within 0.3 min of that for the fortification extract.

H. SAFETY INFORMATION AND PRECAUTIONS

- 1. Required Protective Equipment Safety glasses, plastic gloves, labcoat.
- 2. Hazards

Procedure Step	Hazard	Recommended Safe Procedures
Acetonitrile, hexane, dichloromethane, ethyl acetate	These solvents are corrosive and flammable. Vapors can cause severe skin, eye, and respiratory irritation.	Work should be performed in an efficient fume hood away from heat-generating devices.
Sodium hydroxide	Caustic; may cause irreversible skin and eye damage.	Wear the required protective equipment. Avoid spillage or spattering.

FOOD Salety and Ins	spection Se	rvice, Office d	or Public Health Science
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3. Disposal Procedures			
Procedure Step		Hazard	Recommended Safe Procedures
Acids and Bases		See Above	Neutralize (pH 5-8) and flush into waste sink with large amounts of water if waste meets Federal/State/Local regulations.
Organic Solvent Mixtu	res	See Above	Segregate chlorinated from nonchlorinated solvents until disposal by contract or in-house specialist.

I. QUALITY ASSURANCE PLAN

1. Performance Standard

- a. For screening sets:
 - i. No false negatives at 0.5 ppb for DES and 1.0 ppb for Zeranol.
 - ii. No false positives for the negative control.
- b. For quantitative sets:

Analyte	Analytical Range	Acceptable Recovery	Acceptable Repeatability % CV
DES	$\geq 0.5 \text{ ppb}$	80 - 130	≤ 20
Zeranol	\geq 1.0 ppb	78 - 120	≤ 20

The Measurement Uncertainty and Method Detection Limit should be recalculated yearly or whenever a change that affects method accuracy, precision, or sensitivity occurs.

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2. Critical Control Points and Specifications

Re	ecord	Acceptable Control
a.	Sample weight	$5.0 \text{ g} \pm 0.1 \text{ g}$
b.	Hydrolysis pH	4.5 ± 0.25
c.	ß-Glucuronidase	100 μL
d.	Incubation	35 - 37 °C
e.	Evaporation to dryness	Remove all acetonitrile
f.	Sodium Hydroxide	1.5 mL of 2N

3. Readiness To Perform (FSIS Training Plan)

a. Familiarization

- i. Phase I: Standards. External standard curve is not applicable for this method. Proceed to Phase II.
- ii. Phase II: Analyst's self-fortified samples. Duplicate recoveries at the fortified levels as listed below for three consecutive sets including:
 - (a) 1 set of liver fortified with DES and zeranol.
 - (b) 1 set of liver fortified with DES.
 - (c) 1 set of muscle fortified with DES and zeranol.

DES (ppb)	Zeranol (ppb)
0	0
0.5	1.0
1.0	2.0
2.0	4.0

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- iii. Phase III: Check samples for analyst accreditation.
 - (a) 13 liver samples, including:
 - 1) 5 samples fortified with DES (between 0 and 2.0 ppb).
 - 2) 8 samples fortified with DES and zeranol.
 - (b) 8 muscle samples fortified with DES and zeranol.
 - (c) Report analytical findings to the Quality Assurance Manager (QAM)

Notification from QAM is required to commence official analysis.

- b. Acceptability criteria.
 - i. Refer to section I.1.a. above for screening sets.
 - ii. Refer to section I.1.b. above for quantitative sets.
- 2. Intralaboratory Check Samples
 - a. System, minimum contents.
 - i. Frequency: One per week by each analyst as samples analyzed.
 - ii. Records to be maintained for:
 - (a) All replicate findings.
 - (b) Appropriate charts.
 - (c) Coefficient of correlation of all internal standard curves.
 - b. Acceptability criteria.

Refer to Section I.1 above.

If unacceptable values are obtained, then:

- i. Stop all official analyses by that analyst.
- ii. Take corrective action.
- 5. Sample Acceptability and Stability
 - a. Matrices: Liver or muscle.
 - b. Sample receipt size: Varied; enough to obtain matrix required for all quantitative and confirmation tests and reserve sample.
 - c. Condition upon receipt: Frozen.
 - d. Sample storage:

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- i. Time: \leq 6 months.
- ii. Condition: Frozen.

6. Sample Set

- a. 12 units as determined by DuPont Prep I apparatus.
- b. Screen set will include:
 - i. 2 fortifications at action level,
 - i. a blank, and
 - ii. up to 9 samples.
- c. Quantitation set will include:
 - i. 4-point fortification curve (as described in section F.1.f.),
 - ii. an analyst recovery and
 - iii. up to 7 samples.

7. Sensitivity

Minimum proficiency level (MPL): 0.5 ppb DES, 1.0 ppb Zeranol

J. WORKSHEET

The worksheet on the following page, DES and Zeranol, can be removed for photocopying.

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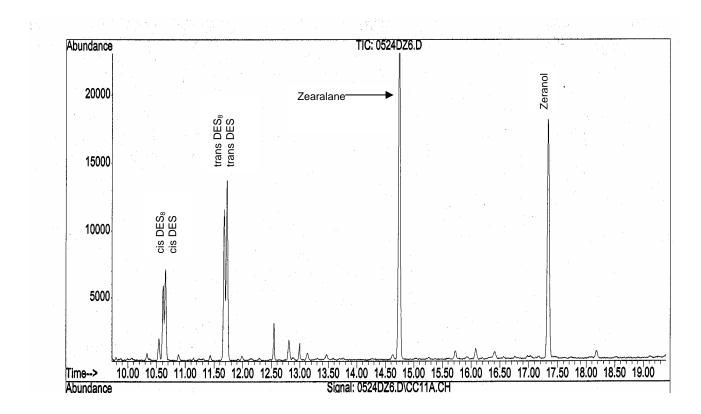
REMARKS:

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K. APPENDIX

1. Chromatogram



1 ppb Standard of DES, Zearalane, and Zeranol.

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Title: Determination and Confirmation of Diethylstilbestrol (DES) and Zeranol by GC/MS		
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2. References

- a. Determination of Diethylstilbestrol and Zeranol in Bovine Liver, Extraction, Screening, Quantitation, and Confirmation. Prepared by Covey, Silvestre, and Henion, Cornell University, Ithaca, NY for USDA, October 15, 1985.
- b. Quantitation of Zeranol and Diethylstilbestrol in Beef Liver, Kidney, and Muscle by GC-MS. Agriculture Canada, Animal Pathology Laboratory, Saskatoon, Saskatchewan, HOR-DP0I.TXT-September 1986.
- c. Method Extension Study on Ovine Tissue for Taleranol, DES, and Zeranol. September, 1988.

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