

EARTH SYSTEM RESEARCH LABORATORY

Ozone Destruction and Recovery

The Role of Laboratory Studies

- A) A General Perspective
- B) The Cl₂O₂ Controversy and Recent CSD Lab Studies

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ESRL Theme Presentation: Ozone Recovery Dec. 4, 2008

What do we mean by "Laboratory Studies" ?

The study of elementary chemical processes relevant to Earth's atmosphere under well-controlled conditions

Objective: Quantify a species

- (1) Reactivity (OH, CI, O_3 , HO_2 , NO_3 ,...)
- (2) Atmospheric lifetime
- (3) Radiative properties

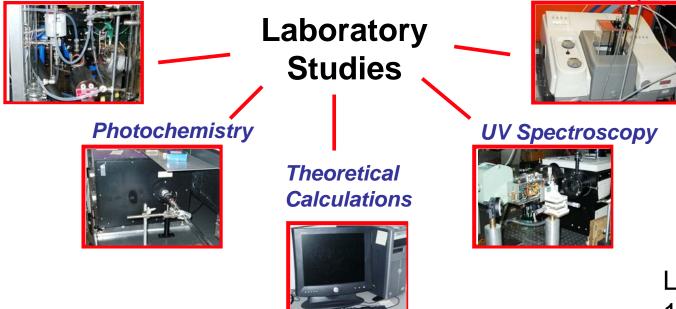
Reaction Kinetics

(4) Environmental Impact

Products

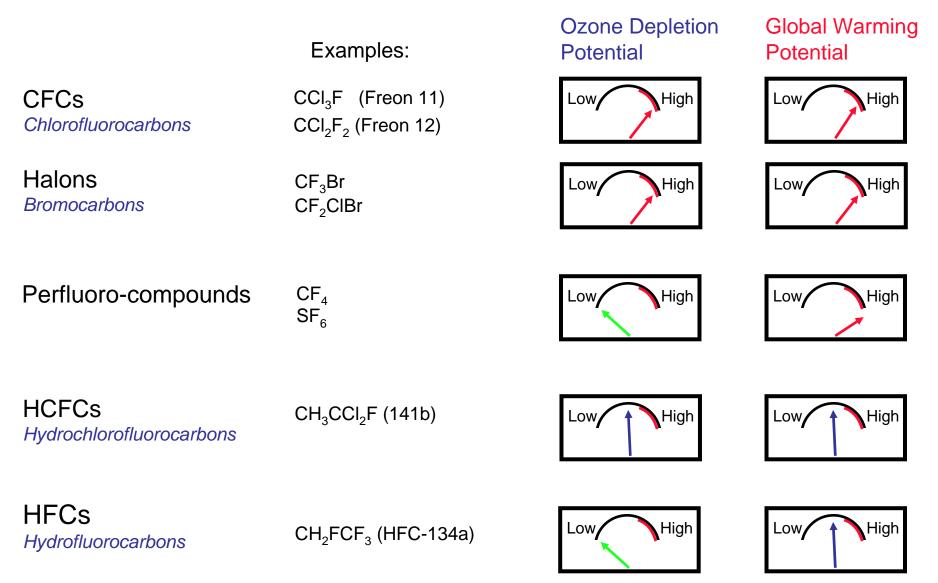
Rates and Mechanisms seconds - 1000s years Radiative Efficiency Air Quality Climate-chemistry coupling Ozone Depletion Potential (ODP) Global Warming Potential (GWP)

Infrared Spectroscopy

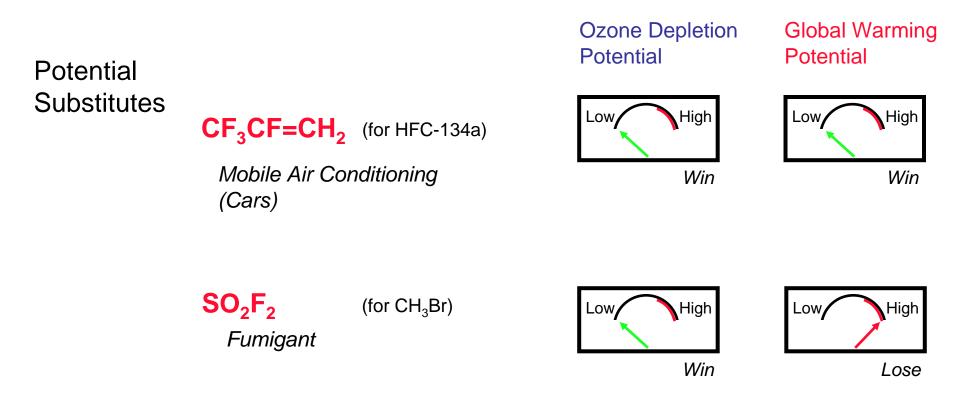


Labs: 1A202, 1A204 Area

Classes of Compounds Studied and Their Impact on Ozone Recovery and Climate



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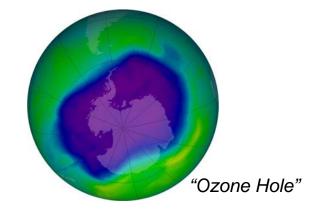


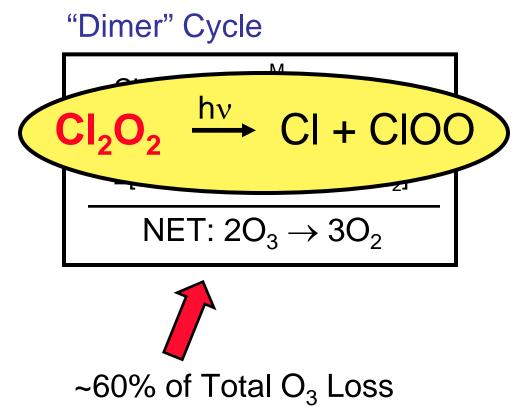
Continuing Area of Research within CSD

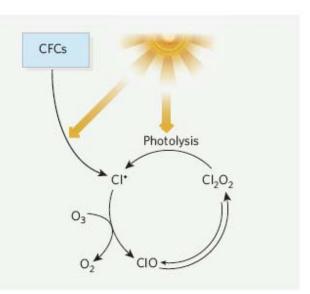
Lab Results Informed Decisions { Commercial Sector Policy (Protocols)

The Cl₂O₂ Controversy and Recent CSD Lab Studies

Chlorine chemistry plays a critical role in Polar Ozone Destruction

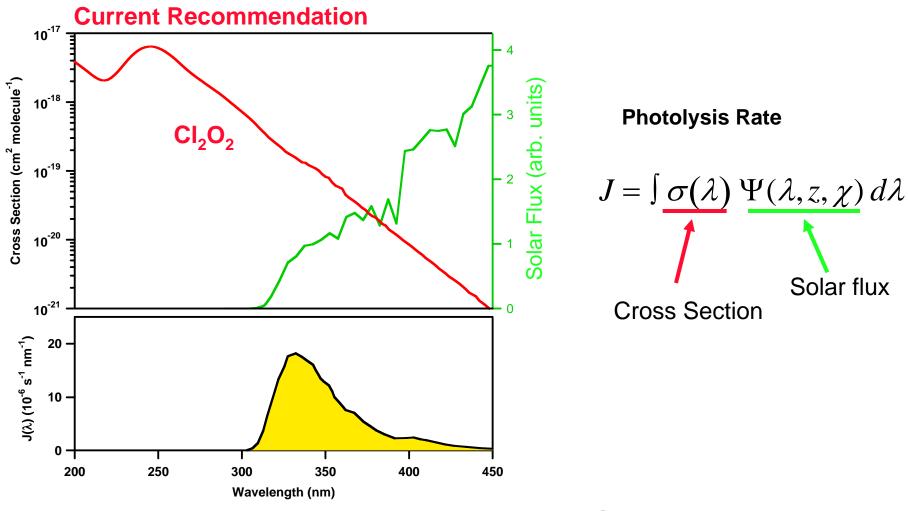






CIO + BrO Cycle: ~ 30%

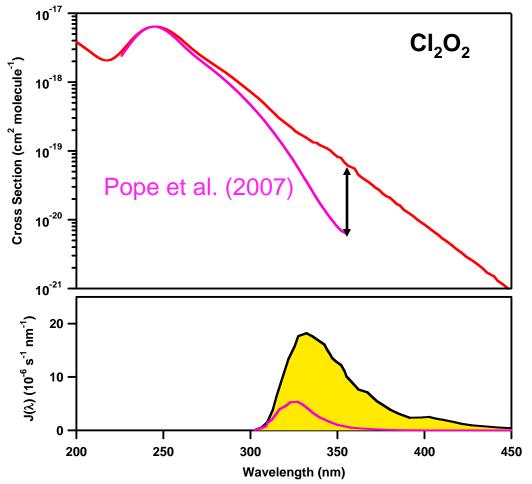
The atmospheric photolysis of Cl₂O₂



Important λ region: 310 - 400 nm

Input for Current Atmospheric Models

A Recently Reported Lab Study of Cl₂O₂ Creates Unrest !



Large Discrepancy with Currently Recommended Values !

Is it important ? YES !!

BIG IMPLICATIONS:

Much Slower Cl₂O₂ photolysis

Poor agreement between observations and models O_3 Loss and Rates [CIO] and [CI₂O₂]





nature

Vol 449|27 September 2007

NEWS

Chemists poke holes in ozone theory

ATMOSPHERIC SCIENCE

Revisiting Ozone Depletion

Marc von Hobe

21 DECEMBER 2007 VOL 318 SCIENCE www.sciencemag.org **Fublished by AAAS**

New laboratory data imply unknown mechanisms in the formation of the ozone hole, but it is too soon to throw out the old paradigms.

Public Awareness

Why is it so hard to get the Cl₂O₂ spectrum Correct?

- * Difficult to prepare bulk samples of Cl₂O₂
- * Small cross sections in λ range of interest
- * Impurities: Spectral Interferences (Cl₂)

New CSD Lab Study: Pulsed UV Laser Photolysis of Cl_2O Objective: Reduce uncertainties in $\sigma(\lambda)$ and spectrum shape

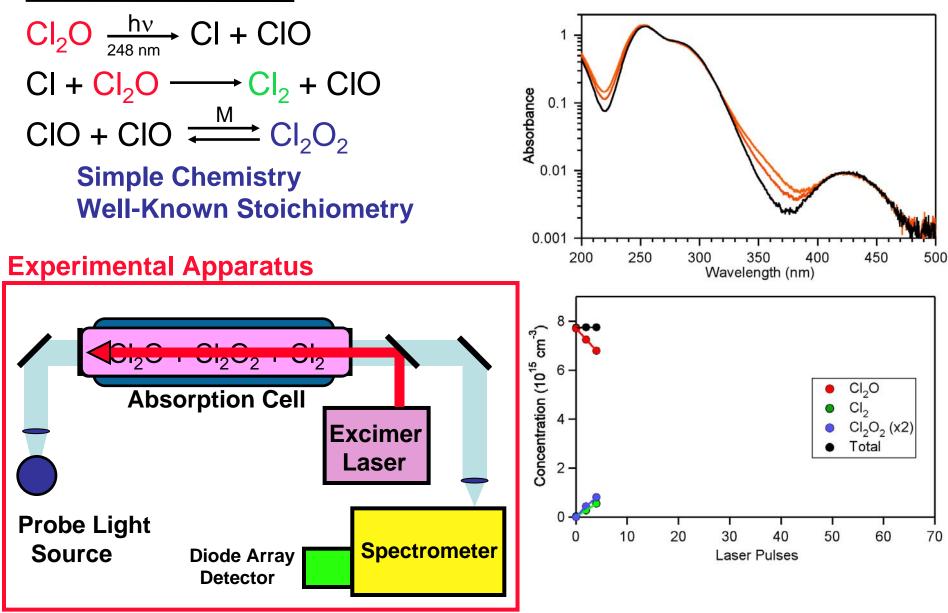


New Approach

- Minimizes uncertainties associated with Cl₂ spectral interference
 A problem in most of the previous studies
- Direct measurements over the critically important wavelength range
 Not achieved in many previous studies

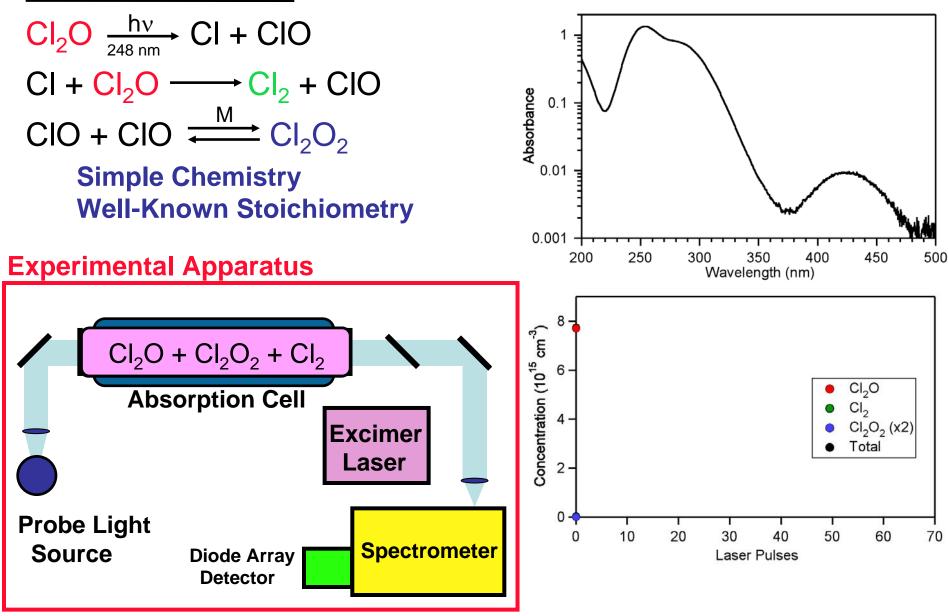
Pulsed UV Laser Photolysis of Cl₂O

Reaction Mechanism

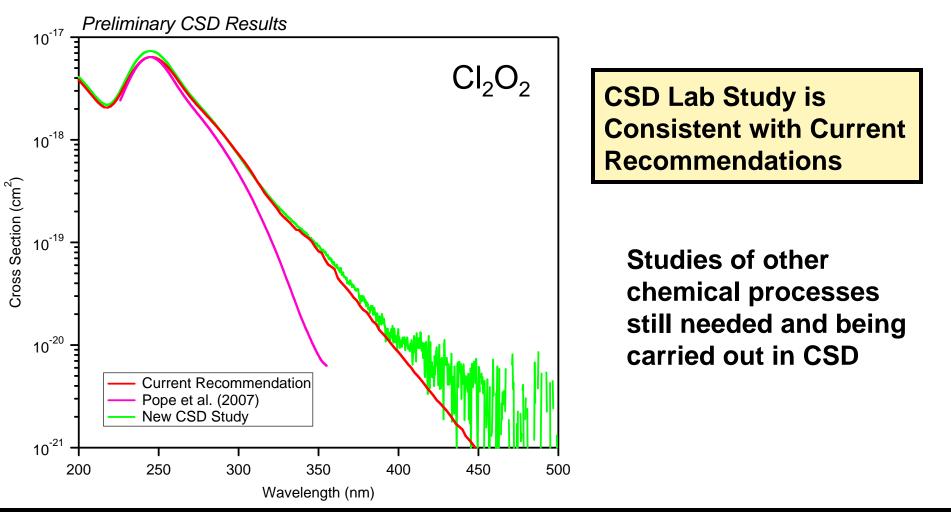


Pulsed UV Laser Photolysis of Cl₂O

Reaction Mechanism



WHAT is the BOTTOM LINE ?



- Laboratory studies play an important role in developing a more thorough understanding of atmospheric processes
- ESRL laboratories address key atmospheric issues